



Natural polymers as stability modifiers of suspensions containing carbonaceous materials obtained by microwave-assisted chemical activation of waste cornelian cherry stones

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ABSTRACT

The cornelian cherry (*Cornus mas L.*) stones were used as a precursor for the preparation of activated biocarbons for cosmetic applications. The microwave-assisted chemical activation with the use of phosphoric acid and potassium carbonate was applied. The resulting carbonaceous materials (CS_P and CS_K) are characterized by a well-developed specific surface area (600 – 800 m²/g) and micro/mesoporous structure with a mean pore diameter of about 2 nm. Their adsorption properties towards hydrolyzed collagen and phytokeatin were checked within the pH range 3–10, and the stability of protein-containing and protein-free suspensions was determined spectrophotometrically. To explain the adsorption and stability mechanisms in the examined systems the solid surface charge and electrokinetic measurements were additionally performed. The surface and electrokinetic properties of both carbon adsorbents were found to be quite similar, e.g. p*H*_{pzc} points of 6.4 and 8.2, and the p*H*_{iep} values of 3.5 and 3.6, for samples CS_P and CS_K, respectively. It was shown that the greatest amounts of both proteins were adsorbed in the case of H₃PO₄-activated biocarbon (587 mg/g at pH 6 for collagen and 585 mg/g at pH 3 for phytokeatin, so very close to their pI values of 5.4 and 3.5, respectively). These systems also exhibit the highest stability among all tested suspensions, as indicated by the smallest aggregate sizes of 23.8 and 21.4 μm, respectively. Such behaviour is highly desirable for cosmetic formulations containing activated biocarbon with adsorbed layers of proteins acting as active agents.

1. Introduction

The Cornelian cherry, or *Cornus mas L.*, is a valuable fruit-bearing plant from the Cornaceae family. It is a deciduous, small shrub or tree reaching a height of 2–5 (up to 9 m) meters. The fruits are drupes of dark-red, sometimes light-red or yellowish-pink color, mainly elliptical or cylindrical, sometimes pear-shaped, ranging from 10 to 15 to 35 mm in length, smooth or slightly granular, sweet-tart in taste. The stone is elliptical or spindle-shaped, almost smooth. All parts of the plant have medicinal properties. Cornelian cherry raw material has antiscorbutic, antidiabetic, antipyretic, anti-inflammatory, bactericidal, general

strengthening, choleric, and diuretic effects. Fresh and dried fruits are used for anemia, inflammatory diseases of the gastrointestinal tract, deficiency of vitamins C and E as well as metabolic disorders. In Georgia, a fixative drug “Schinpani” (dense extract of wild cornelian cherry and pear fruits) has been created, which has undergone clinical trials and received approval for use. The drug made from cornelian cherry pulp in the experiment accelerated protein digestion and showed bactericidal properties against microorganisms of the dysentery group [1–3].

Waste material from cornelian cherry processing, such as stones, can be used to production of activated biocarbons. In this way, such a type of biomass will be utilized and the resulting pyrolysis product will be

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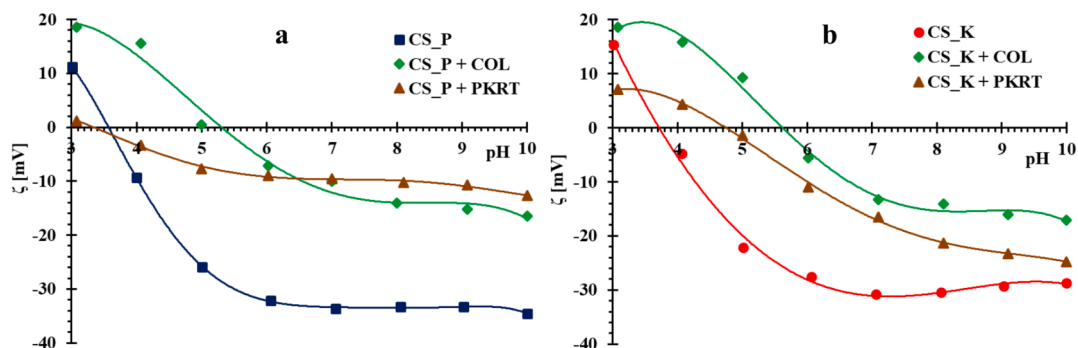


Fig. 13. Zeta potential of CS_P (a) and CS_K (b) particles dispersed in an aqueous solution as a function of solution pH in the absence and presence of proteins.

biocarbon the maximal adsorbed amounts were 585 mg/g for phyto-keratin and 587 mg/g for collagen, respectively. In the analogous systems in the case of CS_K material, the adsorption levels are: 409 and 462 mg/g, respectively. The stability of CS_P suspensions containing both biopolymers is relatively high, when their adsorption reaches considerable levels, specified below. The decrease in suspension stability is accompanied by an increase in the size of the aggregates formed. The densely packed adsorption layers of protein molecules assuming their native conformation guarantee effective steric repulsion between solid particles dispersed in an aqueous solution. The surface charge density and zeta potential values determined for activated biocarbons systems confirmed the adsorption of proteins and indicated the specific conformation of their macromolecules in the surface layer. In the course of the conducted research, it was proven that systems with optimal composition for cosmetic applications were obtained, i.e. those that showed significant adsorption of the active substance (protein) at pH suitable for skin and hair care and were very stable. The best in this aspect turned out to be CS_P activated biocarbon with adsorbed collagen at pH 6. It is worth mentioning that we have already successfully attempted to introduce the tested activated biocarbons modified with hydrolyzed COL into cosmetic products for body and hair care, such as solid shampoo and bar soap (in cooperation with companies “SPA-Vita product” and ZELENAHA).

CRediT authorship contribution statement

Małgorzata Wiśniewska: Conceptualization, Methodology, Validation, Formal analysis, Resources, Writing – original draft, Writing – review & editing, Supervision. **Teresa Urban:** Methodology, Investigation, Data curation, Writing – original draft, Writing – review & editing. **Iwona Ostolska:** Validation, Investigation, Writing – original draft, Visualization. **Karina Tokarska:** Investigation, Resources, Data curation. **Victoria Paientko:** Formal analysis, Investigation, Resources, Visualization. **Alla Kustovska:** Investigation, Data curation, Writing – original draft. **Vita Vedmedenko:** Investigation, Data curation. **Natalia Kurinna:** Investigation, Data curation. **Piotr Nowicki:** Conceptualization, Methodology, Validation, Formal analysis, Investigation, Resources, Writing – original draft, Writing – review & editing, Supervision.

Declaration of competing interest

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests: The authors Vita Vedmedenko and Natalia Kurinna are sole entrepreneurs employed by the companies “SPA – Vita product” and ZELENAHA. The remaining Authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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Data availability

Data will be made available on request.

References

- [1] A.V. Liamicheva, S.V. Klymenko, Introduction of Cornaceae Dumort. family species in Ukraine and prospects of their usage, *Plant Introd.* 2 (1999) 32–36. <http://jnas.nbuv.gov.ua/article/UJRN-0001036374> [In Ukrainian].
- [2] A.V. Kustovska, Rodina Cornaceae (Dumort.) Dumort. v Ukrayini (sistema, biologichni osoblivosti, narodnogospodarske znachennya). – Dis. na zdobuttya nauk.stupenya kand..biol.nauk. – K., NBS im.M.M. Gryshka NANU, (2002) [In Ukrainian].
- [3] R. Spychaj, A.Z. Kucharska, A. Szumny, D. Przybylska, E. Pejcz, N. Piórecki, Potential valorization of Cornelian cherry (Cornus mas L.) stones: Roasting and extraction of bioactive and volatile compounds, *Food Chem.* 358 (2021) 129802, <https://doi.org/10.1016/j.foodchem.2021.129802>.
- [4] F.O. Erdogan, Comparative study of sunset yellow dye adsorption onto cornelian cherry stones-based activated carbon and carbon nanotubes, *Bulgarian Chem. Communicat.* 50 (2018) 592–601.
- [5] E. Demirbas, M. Kobya, E. Senturk, T. Ozkan, Adsorption kinetics for the removal of chromium (VI) from aqueous solutions on the activated carbons prepared from agricultural wastes, *Water SA* 30 (2004) 533–539.
- [6] M. Raininga, A. Mudgal, V.K. Patel, J. Patel, M.K. Sinha, Modification of activated carbon-based adsorbent for removal of industrial dyes and heavy metals: A review, *Mat. Today: Proceed.* 77 (2023) 286–294, <https://doi.org/10.1016/j.matpr.2022.11.358>.
- [7] S. Wong, N. Ngadi, I.M. Inuwa, O. Hassan, Recent advances in applications of activated carbon from biowaste for wastewater treatment: a short review, *J. Cleaner Prod.* 175 (2018) 361–375, <https://doi.org/10.1016/j.jclepro.2017.12.059>.
- [8] P. Hadi, M. Xu, C. Ning, C.S.K. Lin, G. McKay, A critical review on preparation, characterization and utilization of sludge-derived activated carbons for wastewater treatment, *Chem. Eng. J.* 260 (2015) 895–906, <https://doi.org/10.1016/j.cej.2014.08.088>.
- [9] H.R. Lotfy, H. Roubik, Water purification using activated carbon prepared from agriculture waste—overview of a recent development, *Biomass Conv. Biorefin.* 13 (2023) 15577–15590, <https://doi.org/10.1007/s13399-021-01618-3>.
- [10] F. Sher, K. Hanif, A. Rafey, U. Khalid, A. Zafar, M. Ameen, E.C. Lima, Removal of micropollutants from municipal wastewater using different types of activated carbons, *J. Environ. Manag.* 278 (2021) 111302, <https://doi.org/10.1016/j.jenvman.2020.111302>.
- [11] G. León, A.M. Hidalgo, A. Martínez, M.A. Guzmán, B. Miguel, Methylparaben adsorption onto activated carbon and activated olive stones: comparative analysis of efficiency equilibrium, kinetics and effect of graphene-based nanomaterials addition, *Applied Sci.* 13 (2023) 9147, <https://doi.org/10.3390/app13169147>.
- [12] T.B. Devi, D. Mohanta, M. Ahmaruzzaman, Biomass derived activated carbon loaded silver nanoparticles: an effective nanocomposites for enhanced solar photocatalysis and antimicrobial activities, *J. Ind. Eng. Chem.* 76 (2019) 160–172, <https://doi.org/10.1016/j.jiec.2019.03.032>.
- [13] H. Hammani, F. Laghrib, A. Farahi, S. Lahrich, T. El Ouafy, A. Aboulkas, K. El Harfi, M.A. El Mhammedi, Preparation of activated carbon from date stones as a catalyst to the reactivity of hydroquinone: Application in skin whitening cosmetics samples, *J. Sci. Adv. Mat. Dev.* 4 (2019) 451–458, <https://doi.org/10.1016/j.jsamd.2019.07.003>.
- [14] P.S. Abdullah, J.Y. Lim, N.A. Abdullah, Synthesis and characterization of active biocarbon material for use in cosmetics and personal care products, *Key Eng. Mat.* 841 (2020) 266–272, <https://doi.org/10.4028/www.scientific.net/KEM.841.266>.

- [15] H. Kaur, A. Bansawal, G. Hippargi, G.R. Pophali, Effect of hydrophobicity of pharmaceuticals and personal care products for adsorption on activated carbon: adsorption isotherms, kinetics and mechanism, *Environ. Sci. Pollut. Res.* 25 (2018) 20473–20485, <https://doi.org/10.1007/s11356-017-0054-7>.
- [16] D.S. Ratmelya, J. Reveny, U. Harahap, Test anti-aging activity in a face scrub preparation that contains coffee-grade active charcoal (*Coffea arabica* L.) with the addition of vitamin E, *Sci. Ris. Pharmaceut. Sci.* 5 (2022) 74–82, <https://doi.org/10.15587/2519-4852.2022.265402>.
- [17] P. Agrawal, S. Panda, S.R. Mishra, A Review on Activated Charcoal Toothpaste, *World J. Pharmaceut. Res.* 7 (2018) 478–481, <https://doi.org/10.20959/wjpr20189-12132>.
- [18] F. Mellou, A. Varvaresou, S. Papageorgiou, Renewable sources: applications in personal care formulations, *Int. J. Cosmetic Sci.* 41 (2019) 517–525, <https://doi.org/10.1111/ics.12564>.
- [19] G.S. Luengo, A.L. Fameau, F. Leonforte, A.J. Greaves, Surface science of cosmetic substrates, cleansing actives and formulations, *Adv. Colloid Interf. Sci.* 290 (2021) 102383, <https://doi.org/10.1016/j.cis.2021.102383>.
- [20] D. Venkataramani, A. Tsulaia, S. Amin, Fundamentals and applications of particle stabilized emulsions in cosmetic formulations, *Adv. Colloid Interf. Sci.* 283 (2020) 102234, <https://doi.org/10.1016/j.cis.2020.102234>.
- [21] N. Kania, S. Rio, E. Monflier, A. Ponchel, Cyclodextrins adsorbed onto activated carbons: Preparation, characterization, and effect on the dispersibility of the particles in water, *J. Colloid Interf. Sci.* 371 (2012) 89–100, <https://doi.org/10.1016/j.jcis.2011.11.083>.
- [22] M. Wiśniewska, D. Sternik, P. Nowicki, S. Chibowski, M. Medykowska, M. Gęca, K. Szewczuk-Karpisz, Adsorption, viscosity and thermal behaviour of nanosized proteins with different internal stability immobilised on the surface of mesoporous activated biocarbon obtained from the horsetail herb precursor, *Appl. Nanosci.* 12 (2022) 1323–1336, <https://doi.org/10.1007/s13204-021-01759-x>.
- [23] M. Wiśniewska, P. Nowicki, T. Urban, V.M. Gun'ko, Physicochemical and Adsorption Properties of Nanoporous Activated Biocarbons from Thermochemical Treatment of Horsetail Herb, *ChemPlusChem* 89 (2024) e202400177, <https://doi.org/10.1002/cplu.202400177>.
- [24] K. Szewczuk-Karpisz, M. Wiśniewska, P. Nowicki, P. Oleszczuk, Influence of protein internal stability on its removal mechanism from aqueous solutions using eco-friendly horsetail herb-based engineered biochar, *Chem. Eng. J.* 388 (2020) 124156, <https://doi.org/10.1016/j.cej.2020.124156>.
- [25] A.I. Osman, A. Ayati, P. Krivoschapkin, B. Tanhaei, M. Farghali, P.-S. Yap, A. Abdelhaleem, Coordination-driven innovations in low-energy catalytic processes: Advancing sustainability in chemical production, *Coord. Chem. Rev.* 514 (2024) 215900, <https://doi.org/10.1016/j.ccr.2024.215900>.
- [26] B.-X. Jiang, H. Wang, Y.-T. Zhang, S.-B. Li, Microwave-assisted synthesis of Zr-based metal–organic frameworks and metal-organic cages, *Polyhedron* 243 (2023) 116569, <https://doi.org/10.1016/j.poly.2023.116569>.
- [27] M. Wiśniewska, K. Tokarska, T. Urban, P. Nowicki, A. Wozzuk, Production of activated biocarbons by microwave-assisted chemical activation of hardwood sawdust and their application in the simultaneous removal of polymers of different origins from aqueous systems, *Wood Sci. Technol.* 59 (2025) 15, <https://doi.org/10.1007/s00226-024-01623-5>.
- [28] J. Shi, N. Yan, H. Cui, J. Xu, Y. Liu, S. Zhang, Salt Template Synthesis of Nitrogen and Sulfur Co-Doped Porous Carbons as CO₂ Adsorbents, *ACS Sustainable Chem. Eng.* 7 (2019) 19513–19521, <https://doi.org/10.1021/acssuschemeng.9b04574>.
- [29] J. Shi, H. Cui, J. Xu, N. Yan, Y. Liu, Design and fabrication of hierarchically porous carbon frameworks with Fe₂O₃ cubes as hard template for CO₂ adsorption, *Chem. Eng. J.* 389 (2020) 124459, <https://doi.org/10.1016/j.cej.2020.124459>.
- [30] H. Cui, J. Xu, J. Shi, S. You, Ch. Zhang, N. Yan, Y. Liu, G. Chen, Evaluation of different potassium salts as activators for hierarchically porous carbons and their applications in CO₂ adsorption, *J. Colloid Interf. Sci.* 583 (2021) 40–49, <https://doi.org/10.1016/j.jcis.2020.09.022>.
- [31] H. Cui, J. Xu, J. Shi, N. Yan, Y. Liu, Facile fabrication of nitrogen doped carbon from filter paper for CO₂ adsorption, *Energy* 187 (2019) 115936, <https://doi.org/10.1016/j.energy.2019.115936>.
- [32] B. Jadach, Z. Mielcarek, T. Osmatek, Use of Collagen in Cosmetic Products, *Curr. Iss. Molec. Biol.* 46 (2024) 2043–2070, <https://doi.org/10.3390/cimb46030132>.
- [33] C.L. Burnett, W.F. Bergfeld, D.V. Belsito, R.A. Hill, C.D. Klaassen, D.C. Liebler, J. G. Marks, R.C. Shank, T.J. Slaga, P.W. Snyder, L.J. Gill, B. Heldreth, Safety Assessment of Keratin and Keratin-Derived Ingredients as Used in Cosmetics, *Int. J. Toxicol.* 40 (2021) 36S–51S, <https://doi.org/10.1177/10915818211013019>.
- [34] A. León-López, A. Morales-Peñaloza, V.M. Martínez-Juárez, A. Vargas-Torres, D. I. Zeugolis, G. Aguirre-Álvarez, Hydrolyzed Collagen—Sources and Applications, *Molecules* 24 (2019) 4031, <https://doi.org/10.3390/molecules24224031>.
- [35] P. Nowicki, K. Gruszczynska, T. Urban, M. Wiśniewska, Activated biocarbons obtained from post-fermentation residue as potential adsorbents of organic pollutants from the liquid phase, *Physicochem. Probl. Miner. Process.* 58 (2022) 146357, <https://doi.org/10.37190/ppmp/146357>.
- [36] M.H. Simonian, Spectrophotometric determination of protein concentration, *Curr. Protoc. Cell Biol. Appendix 3B* (2002) 18228395, <https://doi.org/10.1002/0471143030.cba03bs15>.
- [37] I. Ostolska, M. Wiśniewska, Investigation of the colloidal Cr₂O₃ removal possibilities from aqueous solution using the ionic polyamino acid block copolymers, *J. Hazard. Mat.* 290 (2015) 69–77, <https://doi.org/10.1016/j.jhazmat.2015.02.068>.
- [38] M. Wiśniewska, P. Nowicki, K. Szewczuk-Karpisz, M. Gęca, K. Jędruchiewicz, P. Oleszczuk, Simultaneous removal of toxic Pb(II) ions, poly(acrylic acid) and Triton X-100 from their mixed solution using engineered biochars obtained from horsetail herb precursor – Impact of post-activation treatment, *Sep. Purif. Technol.* 276 (2021) 119297, <https://doi.org/10.1016/j.seppur.2021.119297>.
- [39] W. Janusz, Electrical double layer at the metal oxide–electrolyte interface in “interfacial forces and fields: theory and applications”, M. Dekker, New York, vol 85, Chapter 4, (1999).
- [40] H.A. Oshima, simple expression for Henry's function for the retardation effect in electrophoresis of spherical colloidal particles, *J. Colloid Interf. Sci.* 168 (1994) 269–271, <https://doi.org/10.1006/jcis.1994.1419>.
- [41] M. Medykowska, M. Wiśniewska, K. Szewczuk-Karpisz, R. Panek, Study on electrical double layer nanostructure on zeolitic materials' surface in the presence of impurities of different nature, *Appl. Nanosci.* 13 (2023) 6737–6748, <https://doi.org/10.1007/s13204-022-02747-5>.
- [42] S.H. Park, T. Song, T.S. Bae, B.H. Choi, S.R. Park, B.-H. Min, Comparative analysis of collagens extracted from different animal sources for application of cartilage tissue engineering, *Int. J. Precis. Eng. Manuf.* 13 (2012) 2059–2066, <https://doi.org/10.1007/s12541-012-0271-4>.
- [43] S. De Oliveira, G. Miklosic, J. Veziars, S. Grastilleur, T. Coradin, C. Le Visage, J. Guicheux, M. D'Este, C. Hélayr, Optimizing the physical properties of collagen/hyaluronan hydrogels by inhibition of polyionic complexes formation at pH close to the collagen isoelectric point, *Soft Matter* 19 (2023) 9027–9035, <https://doi.org/10.1039/d3sm01330h>.
- [44] C.J. Molloy, J.D. Laskin, Keratin polypeptide expression in mouse epidermis and cultured epidermal cells, *Differentiation* 37 (1988) 86–97, <https://doi.org/10.1111/j.1432-0436.1988.tb00800.x>.
- [45] K. Szewczuk-Karpisz, M. Wiśniewska, Adsorption properties of the albumin – chromium(III) oxide system – effect of solution pH and ionic strength, *Soft Mat.* 12 (2014) 268–276, <https://doi.org/10.1080/1539445X.2014.890940>.
- [46] T. Riaz, R. Zeeshan, F. Zarif, K. Ilyas, N. Muhammad, S.Z. Safi, A. Rahim, S.A. A. Rizvi, I. Ur Rehman, FTIR analysis of natural and synthetic collagen, *Appl. Spectrosc. Rev.* 53 (2018) 703–746, <https://doi.org/10.1080/05704928.2018.1426595>.
- [47] C. Stani, L. Vaccari, E. Mitri, G. Birarda, FTIR investigation of the secondary structure of type I collagen: New insight into the amide III band, *Spectrochim. Acta A* 229 (2020) 118006, <https://doi.org/10.1016/j.saa.2019.118006>.
- [48] S. Saravanan, D.K. Sameera, A. Moorthi, N. Selvamurugan, Chitosan Scaffolds containing Chicken Feather Keratin Nanoparticles for Bone Tissue Engineering, *Int. J. Biol. Macromol.* 62 (2013) 481–486, <https://doi.org/10.1016/j.ijbiomac.2013.09.034>.
- [49] A.V. Dobrynin, M. Rubinstain, Theory of polyelectrolytes in solutions and at surfaces, *Prog. Polym. Sci.* 30 (2005) 1049–1118, <https://doi.org/10.1016/j.progpolymsci.2005.07.006>.
- [50] A.V. Dobrynin, Solutions of charged polymers, in: K. Matyjaszewski, M. Möller (Eds.), *Polymer Science: A Comprehensive Reference*, vol. 1, Elsevier B.V., Amsterdam, 2012, pp. 81–132.
- [51] M. Wiśniewska, A Review of Temperature Influence on Adsorption Mechanism and Conformation of Water Soluble Polymers on the Solid Surface, *J. Disp. Sci. Tech.* 32 (2011) 1605–1623, <https://doi.org/10.1080/01932691.2010.528332>.
- [52] R.L. Esparza, M. Ba, E. Pérez, A.G. Goicoechea, Importance of Molecular Interactions in Colloidal Dispersions, *Adv. Cond. Matter Phys.* 683716 (2015) 1–8, <https://doi.org/10.1155/2015/683716>.